Journal of Organometallic Chemistry, 266 (1984) 33-36 Elsevier Sequoia S.A., Lausanne – Printed in The Netherlands

REACTION OF 1,4-PENTADIYNE WITH DIETHYLAMINOTRIMETHYLSTANNANE

BERND WRACKMEYER * and CHRISTOPH BIHLMAYER

Institut für Anorganische Chemie der Universität München, Meiserstrasse 1, D-8000 München 2 (F.R.G.) (Received December 15th, 1983)

Summary

The reaction between 1,4-pentadiyne and diethylaminotrimethylstannane leads predominantly to (E)-2-diethylamino-4-trimethylstannyl-2,4-pentaenyne (with either a 1/1 or 1/2 mol ratio of 1,4-pentadiyne and diethylaminotrimethylstannane). With 1/2 stoicheiometry 1,1,5-tris(trimethylstannyl)penta-1,2-dien-4-yne is formed as a minor product. ¹³C and ¹¹⁹Sn NMR data are reported.

Introduction

Many alkynylstannanes can be prepared conveniently according to eq. 1. Removal of the dialkylamine, which is readily formed, generally leads to pure compounds [1].

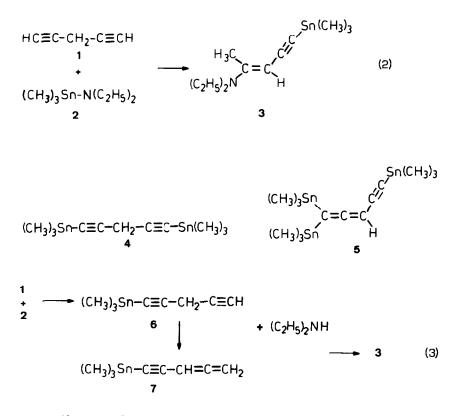
$$n \operatorname{R}^{2} \operatorname{C} = \operatorname{CH} + \left(\operatorname{R}_{2}^{3} \operatorname{N}\right)_{n} \operatorname{Sn} \operatorname{R}_{3 \cdot n}^{1} \to \operatorname{R}_{3 \cdot n}^{1} \operatorname{Sn} \left(\operatorname{C} = \operatorname{CH}\right)_{n} + n \operatorname{R}_{2}^{3} \operatorname{NH}$$
(1)

In previous studies we have shown that alkynylstannanes are interesting precursors of heterocyclic compounds for organoboration reactions [2]. In connection with these studies we attempted the synthesis of 1,4-bis(trimethylstannyl)-1,4-pentadiyne (4), with the results described below.

Results and discussion

Since solutions of 1,4-pentadiyne (1) in tetrahydrofuran (THF) are readily available [3], a reaction similar to eq. 1 appeared to offer an attractive route. However, the reaction takes a different course, as shown in eq. 2. Compound 3 is obtained in high yield in the 1/1 reaction. With excess of the amide 2 (1/2 reaction) a small amount of the allene derivative 5 was also formed (ca. 8%).

We assume that the formation of compound 3 proceeds via an intermediate 6, which rapidly rearranges to the allene derivative 7. The diethylamine which is liberated in the first step probably catalyses this rearrangement and then adds to the allene to give 3 (eq. (3)). This process appears to be stereospecific.



The ¹³C and ¹¹⁹Sn NMR data for the compounds 3 and 4 are shown in Table 1. The assignment of the ¹³C resonances is based upon the magnitude of ${}^{n}J({}^{119}\text{Sn}{}^{-13}\text{C})$ [4] and upon the ¹H coupled ¹³C NMR spectra. The structure of the allene derivative 5 is nicely confirmed by the $\delta({}^{119}\text{Sn})$ values [4,5] and by the coupling constants ${}^{n}J({}^{119}\text{Sn}{}^{-119}\text{Sn})$ (n = 2, 6). The relative position of the substituents at the C=C double bond follows from the $\delta({}^{13}\text{C})$ values and from nuclear Overhauser effect (NOE) difference ¹H NMR spectra [6].

It is concluded that reactions of the type shown in eq. 2 offer a convenient way to new enynes.

Experiments¹

(E)-2-Diethylamino-5-trimethylstannyl-2,4-pentaenyne (3)

Diethylaminotrimethylstannane (2) (3.5 g, 15 m*M*) was added at 20°C under N₂ to a solution of 1,4-pentadiyne (1) (16 m*M*) in 20 ml THF. The mixture was kept for 12 h at 60°C, and its colour changed to orange-red. THF was removed in vacuo and **3** was obtained as a colourless, air-sensitive liquid (3.8 g, 84%, b.p. $86-90^{\circ}C/10^{-1}$ Torr) by fractional distillation. Found: C, 48.04; H, 7.73. C₁₂H₂₃NSn calcd.: C, 48.2; H, 7.6%.

When compound 2 was used in excess (7 g, 30 mM), similar work up gave a fraction boiling between $86-96^{\circ}C/10^{-1}$ Torr, which was found to contain 3 and ca. 8% (NMR) of compound 5, and some unidentified impurities in minor amounts (<3%).

Compound	δ(¹³ C) (ppm)	(H					δ(¹¹⁹ Sn(1))	δ(¹¹⁹ Sn(5))
	C(1)	C(2)	C(3)	C(4)	C(5)	others	(mdd)	(mqq)
	16.6	153.4	74.0	111.6	89.3	43.2 (NCH ₂)	1	- 72.5
	I	(5.0)	(14.2)	(110.0)	(506.7)	[135.0] 12.6 (NCH ₂ CH ₃) [176.1]		
	[127.5,	I	[157.3,	ł	ł	- 8.1 (SnCH ₃) (402.4)		
	6.8]		4.1]			[128.9]		
	79.3	204.2	54.1	105.5	91.6	$- {}^{h}(Sn(1)CH_{3})$	+ 1.7	- 69.8
"J(¹¹⁹ Sn(5)- ¹³ C)	I	1	(15.6)	(102.0)	(484.7)	(350.7 °)	$(105)^2 J$	(52.0) ⁶ J
'(¹¹⁹ Sn(1)- ¹³ C)	(n.o. ^d)	(31.2)	(58.0)	(39.7)	(19.1)	$- h(Sn(5)CH_3)$		
	I	[3.6]	[170.0]	[4.0]	I	(404.0 [°])		

 0 (1⁹Sn) relative to (CH₃)₄Sn external, coupling constants 0 (1¹⁹Sn⁻¹³C) and 0 (1¹⁹Sn⁻¹¹³Sn) in parentheses, coupling constants 0 (1¹³C⁻¹H) in square brackets. ^b Uncertain assignment due to overlap with 13 C(Sn-CH₃) resonance of 3 and spurious signals. ^c From ¹¹⁹Sn⁺¹¹Sn(¹H)-spectra.^d Insufficient signal/noise ratio.

35

TABLE 1

¹³C AND ¹¹⁹Sn NMR DATA " FOR PRODUCTS 3 AND 5

Acknowledgment

Support of this work by the Deutsche Forschungsgemeinschaft and by the Fonds der Chemischen Industrie is gratefully acknowledged.

References

- (a) K. Jones and M.F. Lappert, Proc. Chem. Soc., (1964) 22.
 (b)Yu.I. Derganov, T.F. Gerega and O.S. Dyachkovskaja, Russ. Chem. Revs., 46 (1977) 1132, and ref. cited therein.
- 2 B. Wrackmeyer, Rev. Silicon, Germanium, Tin and Lead Compds., 6 (1982) 75.
- 3 (a) D.A. Ben-Efraim and F. Sondheimer, Tetrahedron, 25 (1969) 2823; (b) P. Jutzi and J. Baumgartner, J. Organomet. Chem., 148 (1978) 257.
- 4 B. Wrackmeyer, Ann. Rep. NMR Spectrosc., in press.
- 5 (a) B. Wrackmeyer, J. Magn. Reson., 39 (1980) 359;
 - (b) J. Organomet. Chem., 205 (1981) 1;
 - (c) B. Wrackmeyer, C. Bihlmayer und M. Schilling, Chem. Ber., 116 (1983) 3182.
- 6 R. Richarz and K. Wüthrich, J. Magn. Reson., 30 (1978) 147.